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### [NAMING IONIC COMPOUNDS WORKSHEET 2](#)

Naming Ionic Compounds Worksheet 2 W 308 Name the following ionic compounds: 1)  $\text{Mo}(\text{SO}_4)_3$  2)  $\text{Ti}(\text{NO}_3)_3$  3)  $\text{NaH}$  4)  $\text{BeSO}_3 \cdot 7\text{H}_2\text{O}$  5)  $\text{Fe}_3(\text{PO}_4)_2$  6)  $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$  7)  $\text{V}(\text{H}_2\text{PO}_4)_4$  8)  $\text{Mn}_3(\text{PO}_4)_7$  9)  $\text{Fe}(\text{CN})_3$  10)  $\text{BaSO}_4$  Write the formulas for the following ionic compounds: 11) vanadium (V) arsenide 12) cobalt (III) nitrite 13) yttrium perchlorate 14) silver phosphide 15) tin (IV) selenide 16) cadmium sulfite 17) nickel (III)...

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### [CHEMICAL BONDING](#)

Lab 2: Naming Ionic Compounds Vocabulary cations positively charged ions anions negatively charged ions cations anions  $\text{Na}^+$   $\text{Ca}^{+2}$   $\text{NH}_4^+$   $\text{Cl}^-$   $\text{NO}_3^-$   $\text{OH}^-$   $\text{Ag}^+$   $\text{Al}^{+3}$   $\text{Mg}^{+2}$   $\text{SO}_4^{2-}$   $\text{PO}_4^{3-}$  ...

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..... 3 SECTION II. INSTRUCTIONS TO BIDDERS  
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